

# Worth County Chemistry and Honor's Chemistry 2017-18

	1 <sup>st</sup> 9 WEEKS	2 <sup>nd</sup> 9 WEEKS	3 <sup>rd</sup> 9 WEEKS	4 <sup>th</sup> 9 WEEKS
<b>Week 1</b>	<ul style="list-style-type: none"> <li>Course introduction</li> <li>Scientific Method</li> <li>Lab Safety</li> <li>Lab procedures</li> </ul>	Bonding (SC2d, SC2e) <ul style="list-style-type: none"> <li>Ionic Formulas and Naming</li> <li>Covalent Formulas and Naming</li> <li>Electronegativity and bonds</li> </ul>	Reactions <ul style="list-style-type: none"> <li>Stoichiometry (SC3d, SC3c)</li> </ul>	Solutions (SC6e) <ul style="list-style-type: none"> <li>Colligative Properties</li> </ul>
<b>Week 2</b>	<ul style="list-style-type: none"> <li>significant figures</li> <li>scientific notation</li> <li>accuracy/precision</li> </ul>	Bonding (SC2d, SC2e) <ul style="list-style-type: none"> <li>Ionic Formulas and Naming</li> <li>Covalent Formulas and Naming</li> <li>Electronegativity and bonds</li> </ul>	Reactions <ul style="list-style-type: none"> <li>Stoichiometry (SC3d, SC3c)</li> </ul>	Acids/Bases (SC6f) <ul style="list-style-type: none"> <li>Properties and uses of common household</li> <li>pH</li> </ul>
<b>Week 3</b>	<ul style="list-style-type: none"> <li>significant figures</li> <li>scientific notation</li> <li>accuracy/precision</li> </ul>	Bonding and Compounds <ul style="list-style-type: none"> <li>Identifying types of bonds (covalent and ionic)</li> <li>Percent Composition</li> </ul>	Reactions <ul style="list-style-type: none"> <li>Stoichiometry (SC3d, SC3c)</li> </ul>	Acids/Bases (SC6g,h) <ul style="list-style-type: none"> <li>Arrhenius</li> <li>Bronsted-Lowry</li> <li>Neutralizations</li> </ul>
<b>Week 4</b>	Characteristics of Matter <ul style="list-style-type: none"> <li>Physical and chemical properties/changes (SC2a, SC2b)</li> <li>Relate function and properties (SC2c)</li> </ul>	Bonding and Compounds (SC3c) <ul style="list-style-type: none"> <li>Percent Composition</li> <li>Empirical/molecular formulas</li> </ul>	Reactions <ul style="list-style-type: none"> <li>Limiting Reactants (SC3e)</li> <li>Percent Yield (SC3c)</li> </ul>	Gas Laws (Sc5b) <ul style="list-style-type: none"> <li>States of Matter</li> <li>Phase Changes—heating curve</li> <li>Flow of energy during phase change</li> </ul>
<b>Week 5</b>	Atomic Structure <ul style="list-style-type: none"> <li>Atomic theory/models (SC1a)</li> <li>Subatomic particles (SC1b,SC1e)</li> <li>Fission/fusion (SC1c)</li> </ul>	Reactions <ul style="list-style-type: none"> <li>Types of reactions (SC3a)</li> </ul>	Reactions <ul style="list-style-type: none"> <li>Limiting Reactants (SC2e)</li> <li>Percent Yield (SC2c)</li> </ul>	Gas Laws (SC5c) <ul style="list-style-type: none"> <li>Ideal and Combined</li> </ul>
<b>Week 6</b>	Atomic Structure <ul style="list-style-type: none"> <li>Moles (SC3c)</li> <li>Relative and average atomic mass (SC1d)</li> </ul>	Reactions (SC3a) <ul style="list-style-type: none"> <li>Predicting &amp; Balancing reactions</li> <li>Activity Series</li> </ul>	Reaction Energy (SC4b,c,d) <ul style="list-style-type: none"> <li>Activation Energy</li> <li>Catalysts</li> <li>LeChatelier's principle</li> </ul>	Thermodynamics <ul style="list-style-type: none"> <li>Heat energy (SC5a)</li> <li>Begin Capstone project</li> </ul>
<b>Week 7</b>	Periodic Trends <ul style="list-style-type: none"> <li>Orbital and electron configuration (SC1g)</li> </ul>	Reactions (SC3a) <ul style="list-style-type: none"> <li>Predicting &amp; Balancing reactions</li> </ul>	Reaction Kinetics <ul style="list-style-type: none"> <li>P, T, concentration, and catalyst effects (SC4a)</li> </ul>	Capstone Project
<b>Week 8</b>	Periodic Trends (SC1f) <ul style="list-style-type: none"> <li>Atomic radius</li> </ul>	Reactions (SC3b)	Solutions (SC6a,b,c,d) <ul style="list-style-type: none"> <li>Solubility factors</li> </ul>	Capstone Project—presentations Review for Final Exam

Common Core Literacy standards will be incorporated throughout the course.

Updated 5/25/17

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	<ul style="list-style-type: none"><li>• Ionic radius</li><li>• Ionization energy</li><li>• Electronegativity</li></ul>	<ul style="list-style-type: none"><li>• Indicators of a chemical reaction</li></ul>	<ul style="list-style-type: none"><li>• Solubility rates</li><li>• Concentration (dilutions)</li></ul>	
<b>Week 9</b>	Bonding (SC2d) <ul style="list-style-type: none"><li>• Types of ions(SC2d)</li><li>• Electronegativity in bonding</li></ul>	Review and Midterm Exam	Solutions <ul style="list-style-type: none"><li>• Molarity</li><li>• Molality</li><li>• Concentration</li></ul>	Review for Final Exam